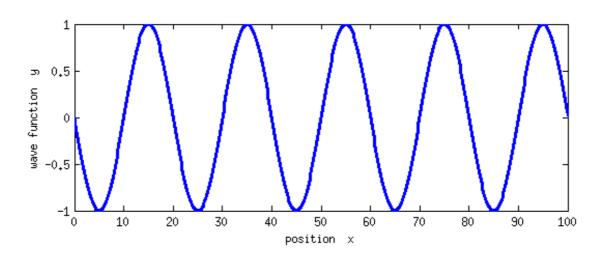
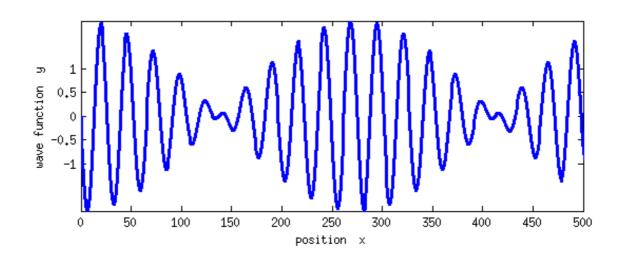
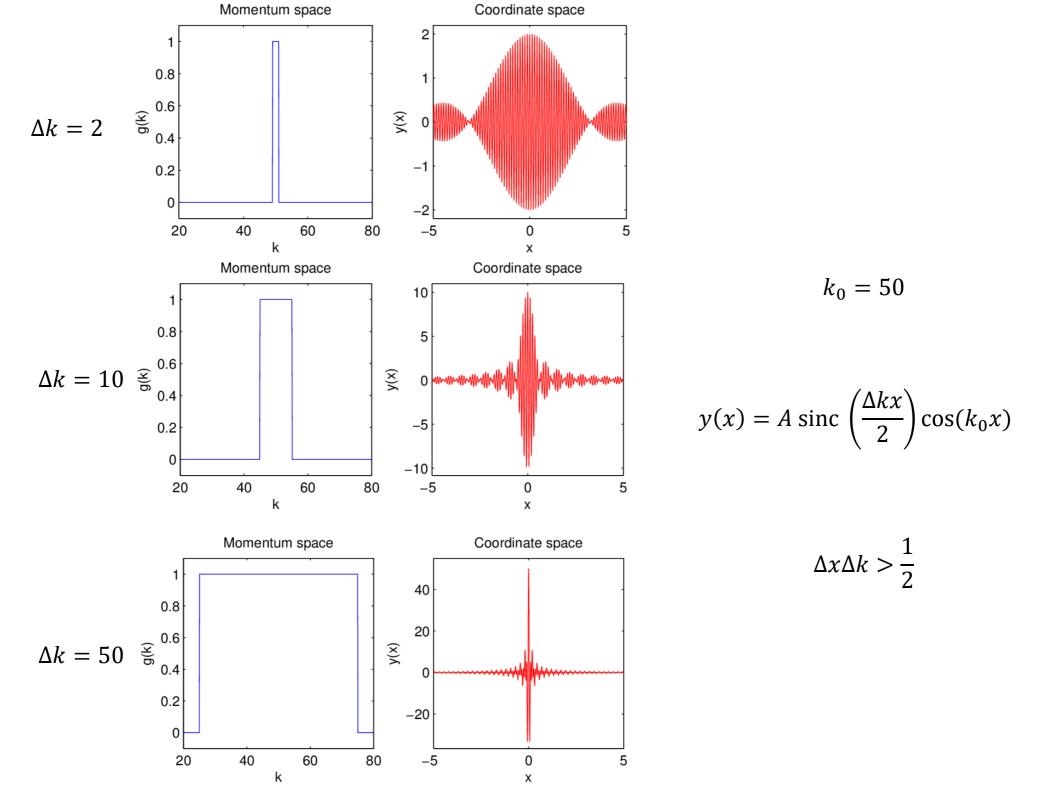
Waves and Fourier Expansion

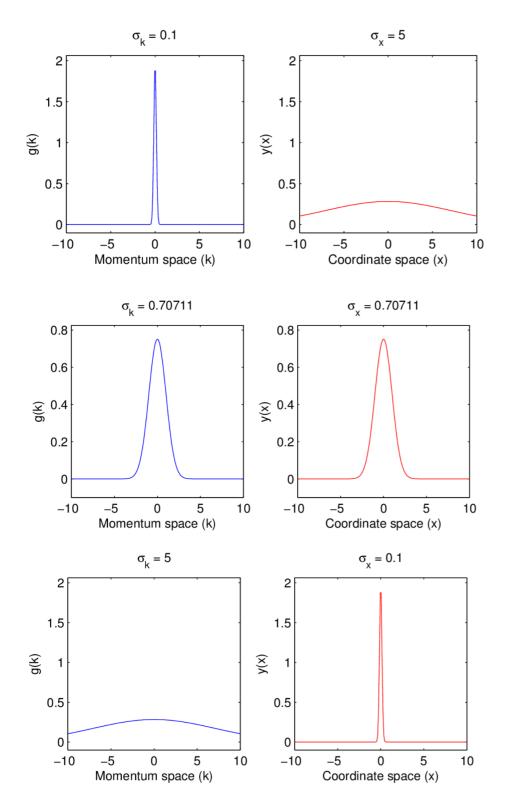
A single travelling sine wave.



An example of beats by adding two sine waves.







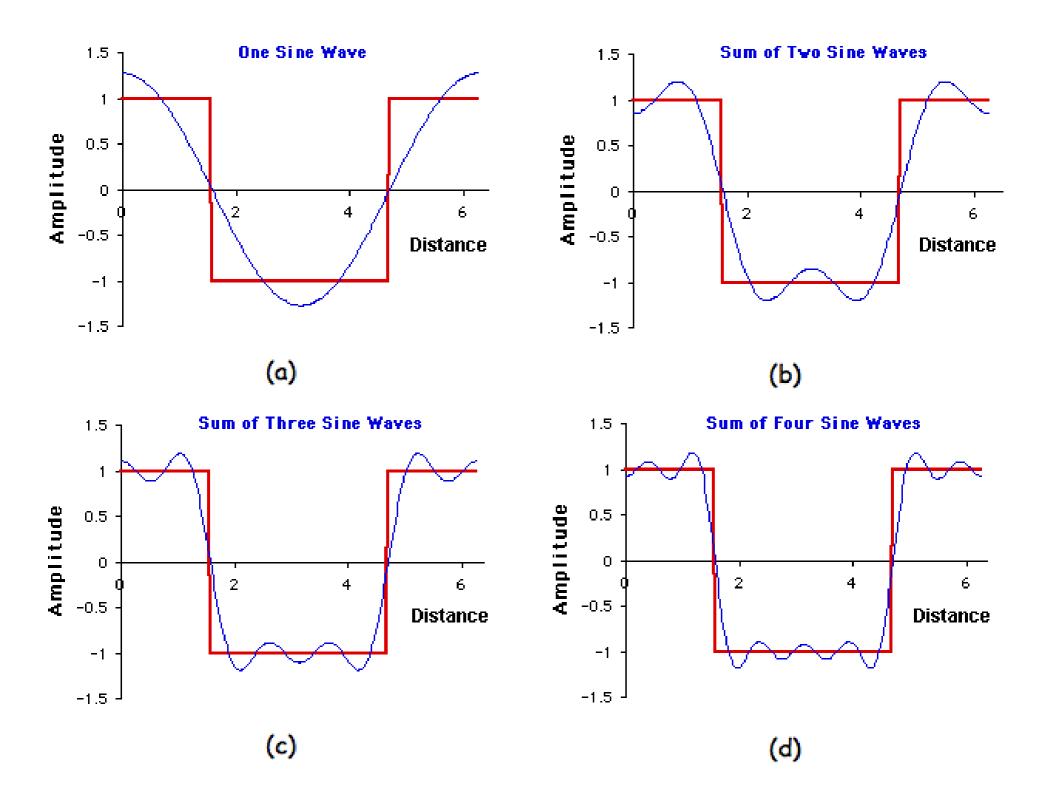
The Fourier transform of a Gaussian is another Gaussian...

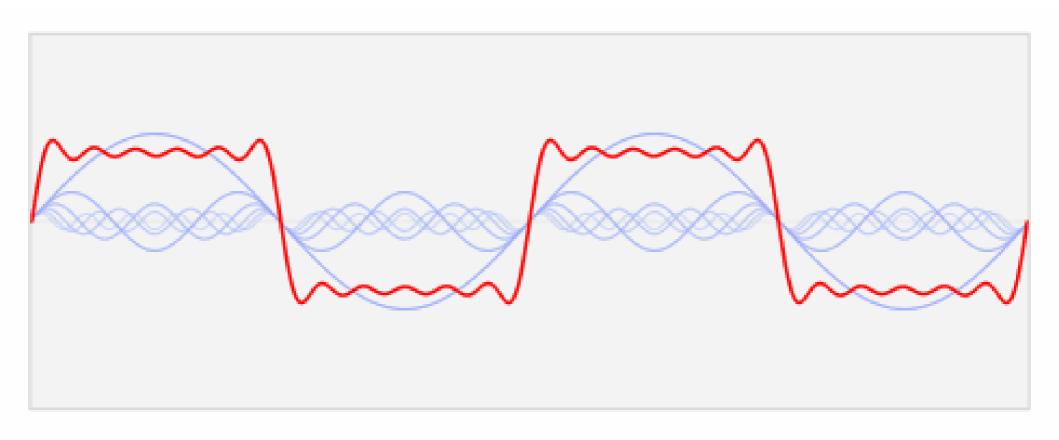
$$g(k) = \frac{1}{\sqrt{2\pi} \sigma_k} \exp(\frac{-k^2}{2\sigma_k^2})$$

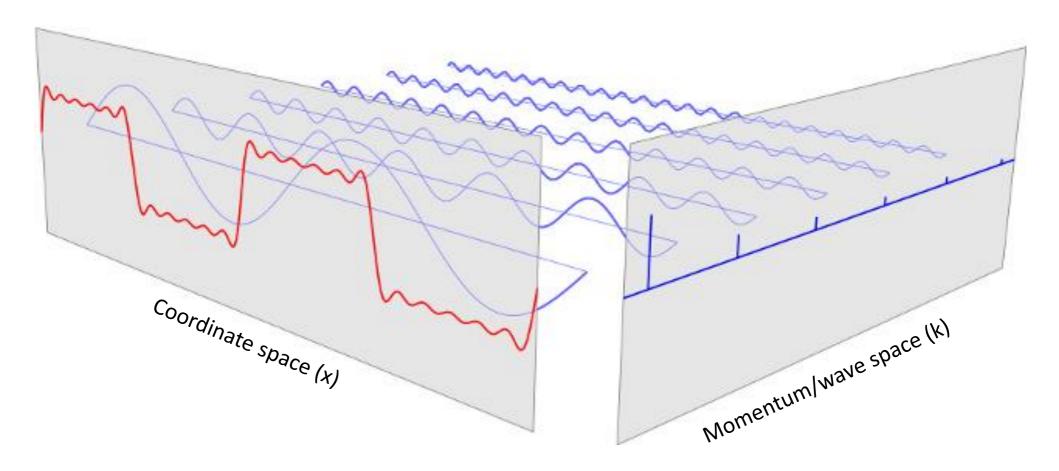
$$y(x) = \frac{1}{\sqrt{2 \pi} \sigma_x} \exp(\frac{-x^2}{2 \sigma_x^2})$$

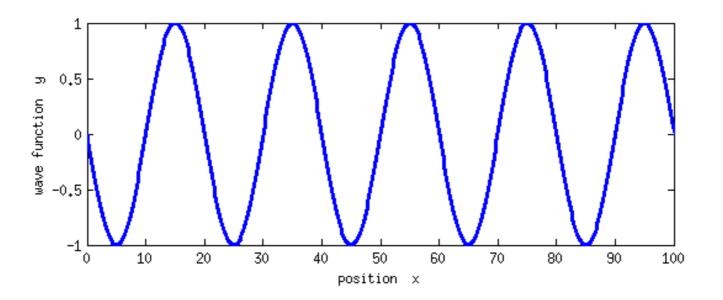
$$\sigma_x \sigma_k = \frac{1}{2}$$

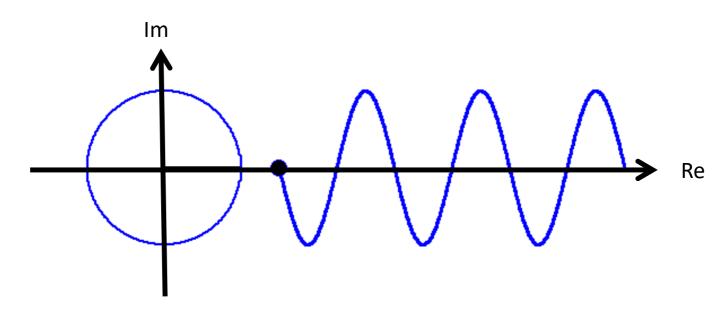
$$\Delta x \Delta k > \frac{1}{2}$$



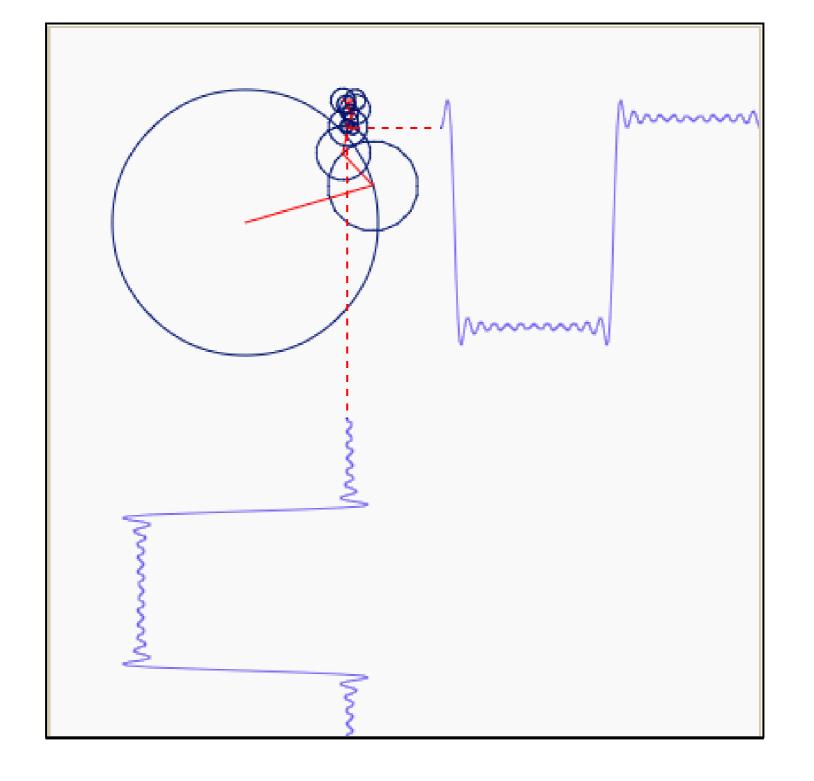








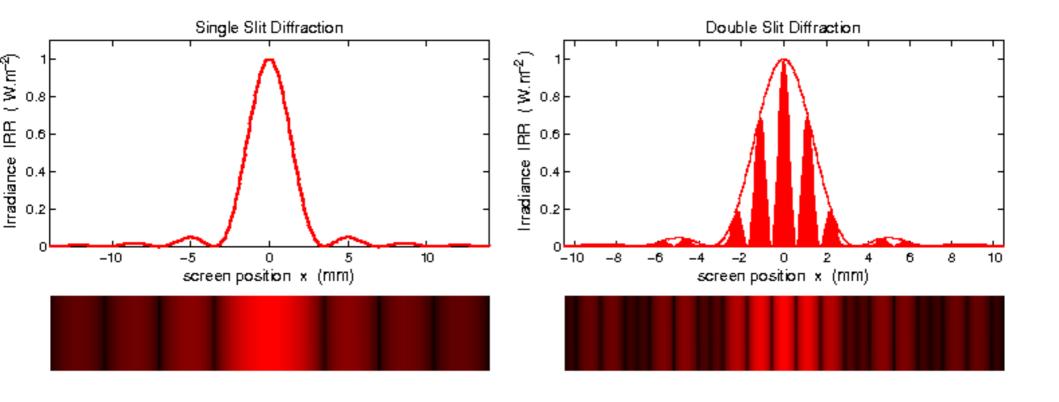
$$\exp(i\theta) = \cos(\theta) + i\sin(\theta)$$

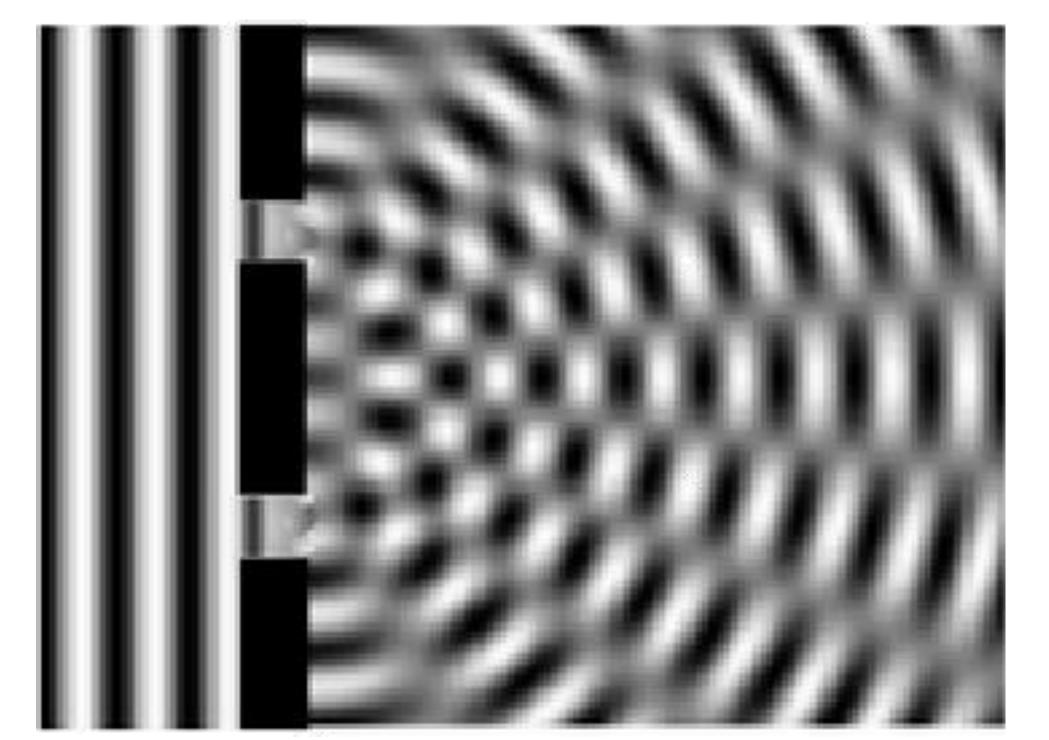


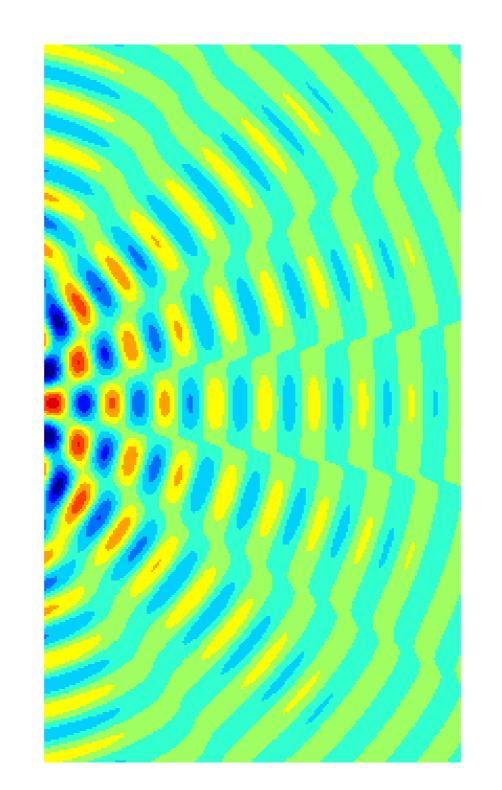
Copenhagen Interpretation of Quantum Mechanics

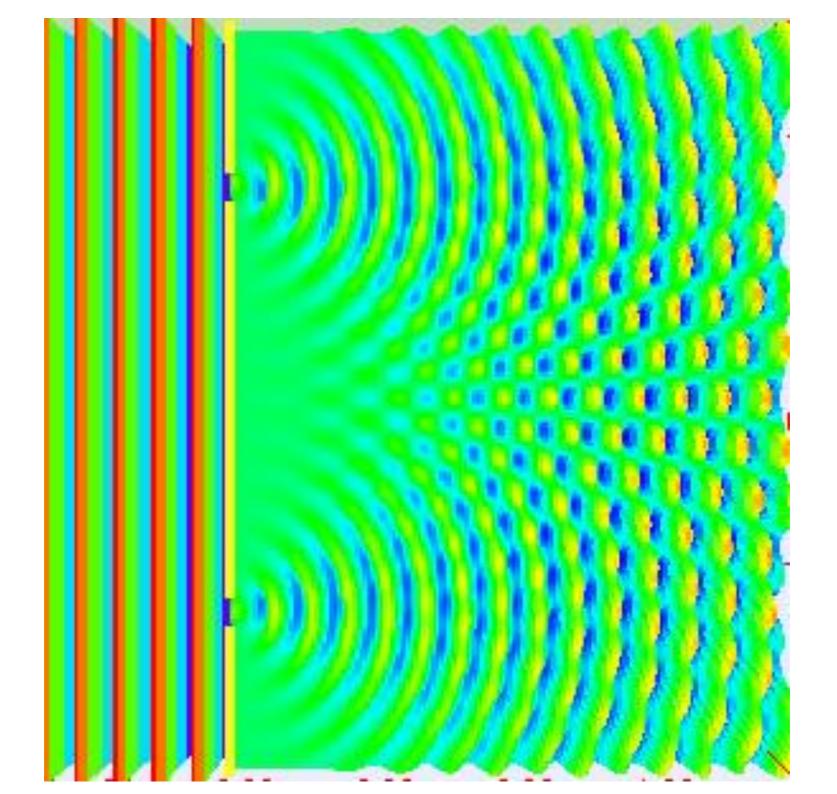
- 1.) A wave function represents the state of a system. It contains all information about anything that can be known about a system before an observation. The wavefunction evolves smoothly in time.
- 2.) Certain properties cannot be known simultaneously for a given system (Heisenberg's Uncertainty Principle). Momentum is meaningless for a perfectly localized particle.
- 3.) During an observation, the system must interact with a laboratory device. When that device makes a measurement, the wavefunction of the system collapses. Collapse of the wavefunction means a wavefunction irreversibly reduces to an eigenfunction corresponding to a particular eigenvalue.
- 4.) The results provided by measuring devices are classical, and must be stated in classical terms.
- 5.) The Born rule: Wavefunctions give a probabilistic description of an experimental outcome.
- 6.) Complementarity rule (Bohr): The wavefunction demonstrates a fundamental waveparticle duality that is a necessary feature of nature.
- 7.) Correspondence rule (Bohr and Heisenberg): When quantum numbers are large, the behaviour of a system reproduces classical predictions.

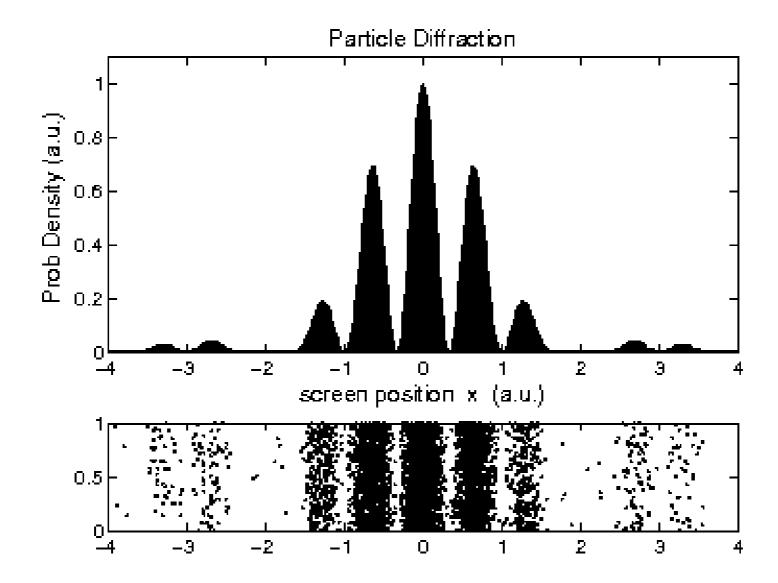
The double-slit experiment for particles



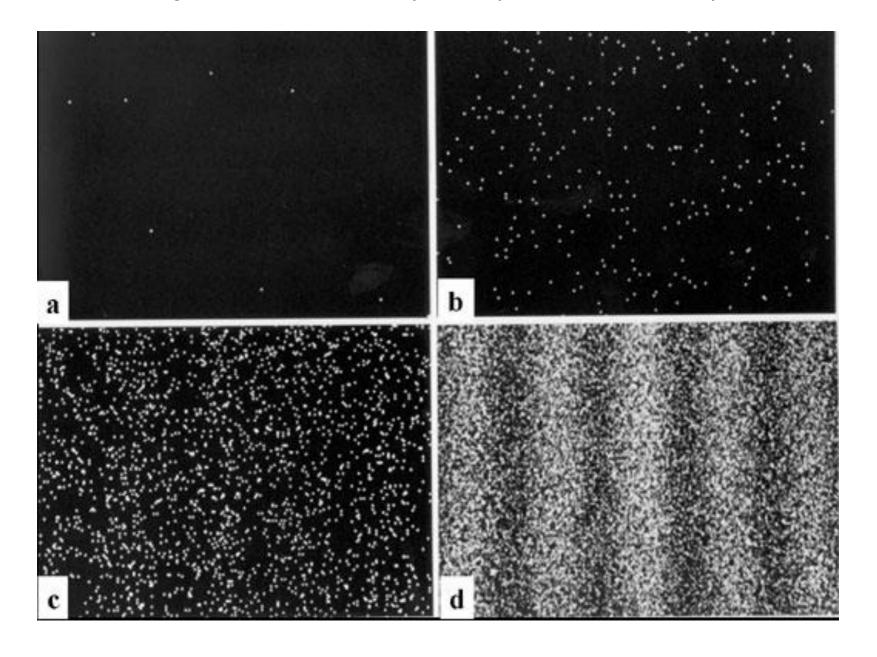








Single electron diffraction pattern (20 minutes duration)



a) 8 electrons

b) 270 electrons

- c) 2000 electrons
- d) 160,000 electrons

