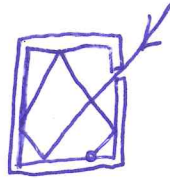


Wave modes for Cavity radiation



$$R(\nu)d\nu = \frac{8\pi}{3} u(\nu)d\nu, \quad u(\nu)d\nu = \frac{1}{V} \int u(\nu)d\nu$$

E_{ox}, E_{oy}, E_{oz} constant ①
 n_x, n_y, n_z integer > 0

$$E_x = E_{ox} \cos\left(\frac{n_x \pi x}{L}\right) \sin\left(\frac{n_y \pi y}{L}\right) \sin\left(\frac{n_z \pi z}{L}\right) \sin(2\pi \nu t)$$

$$E_y = E_{oy} \sin\left(\frac{n_x \pi x}{L}\right) \cos\left(\frac{n_y \pi y}{L}\right) \sin\left(\frac{n_z \pi z}{L}\right) \sin(2\pi \nu t)$$

$$E_z = E_{oz} \sin\left(\frac{n_x \pi x}{L}\right) \sin\left(\frac{n_y \pi y}{L}\right) \cos\left(\frac{n_z \pi z}{L}\right) \sin(2\pi \nu t)$$

$$\frac{\partial^2 E_x}{\partial x^2} + \frac{\partial^2 E_x}{\partial y^2} + \frac{\partial^2 E_x}{\partial z^2} = \frac{1}{c^2} \frac{\partial^2 E_x}{\partial t^2}$$

Any of the components give the condition for standing waves:

~~scribble~~ ~~scribble~~

$$-\left[\left(\frac{n_x \pi}{L}\right)^2 + \left(\frac{n_y \pi}{L}\right)^2 + \left(\frac{n_z \pi}{L}\right)^2\right] E_x = -\frac{(2\pi \nu)^2}{c^2} E_x$$

$$n_r = (n_x^2 + n_y^2 + n_z^2)^{\frac{1}{2}} = \frac{2L\nu}{c}$$

How many wave modes meet this condition?
How to count the modes?

n_x, n_y, n_z integer and positive \rightarrow

n_r is the equation of a sphere!

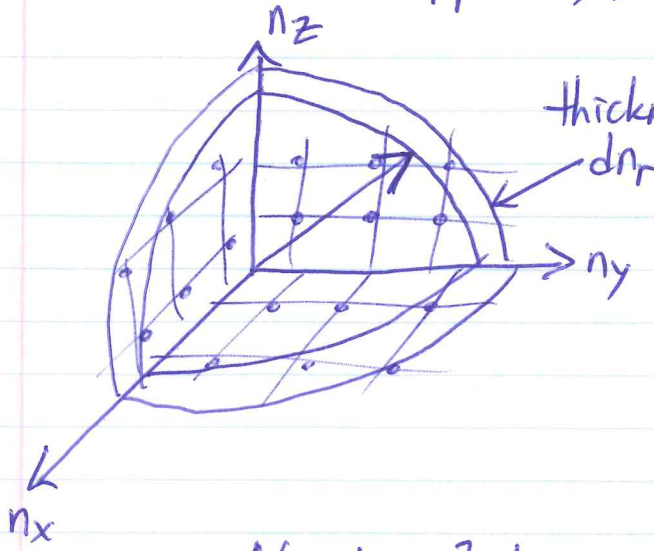
Consider a spherical shell with thickness dn_r

*Note $\langle E \rangle = \langle E \rangle = \bar{E}$ = mean energy

Hibroy

②

$$\vec{n}_r = n_x \hat{x} + n_y \hat{y} + n_z \hat{z}$$



Number of states per unit volume = 1

Positive part of sphere:
 $\frac{V_{r=2}}{8}$ ↑ polarization modes

$$N = \frac{4\pi n_r^2 dn_r}{4}$$

$$\frac{dn_r}{d\nu} = \frac{2L}{c}$$

$$= \pi \left(\frac{4L^2 \nu^2}{c} \right) \left(\frac{2L d\nu}{c} \right)$$

$$N = \frac{8\pi \nu^2 L^3}{c^3} d\nu$$

$$V_{\text{box}} = L^3$$

$$u(\nu) d\nu = \frac{N \langle E \rangle}{V}$$

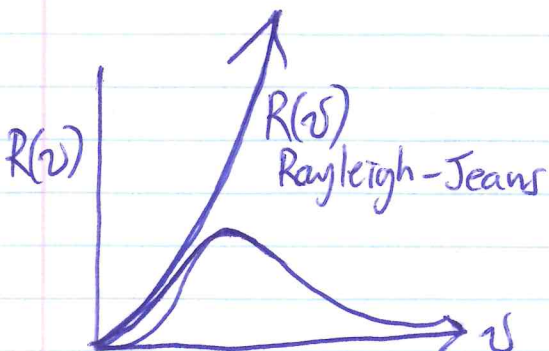
$\langle E \rangle = kT$ from classical physics

~~u(\nu) d\nu = \frac{N \langle E \rangle}{V}~~

$$u(\nu) d\nu = \frac{8\pi \nu^2 kT}{c^3} d\nu$$

$$R(\nu) d\nu = \frac{c}{4} u(\nu) d\nu = \frac{2\pi \nu^2}{c^2} kT d\nu$$

"Rayleigh-Jeans Law"

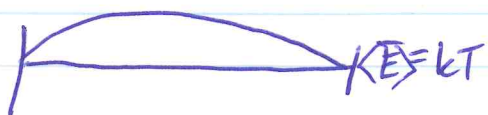


$$R = \int_0^{\infty} R(\nu) d\nu \rightarrow \infty !$$

ULTRAVIOLET CATASTROPHE!

(3)

The problem is with the average energy per wave mode, $kT = \langle E \rangle$



Maxwell-Boltzmann distribution

$$f(E) = \frac{e^{-E/k_b T}}{k_b T}$$



$$k_b = 1.381 \times 10^{-23} \text{ J/K}$$



Maxwell-Boltzmann distribution describes how energy is distributed among a large number of objects (oscillators in BB walls).

$$\langle E \rangle = \frac{\int_0^{\infty} E f(E) dE}{\int_0^{\infty} f(E) dE} = k_b T \quad \text{Planck's solution:}$$

E IS NOT CONTINUOUS

Instead, let $E = nh\nu$ $h = \text{constant}$
 $n = + \text{integer}, n \geq 0$

$$\text{Then } \langle E \rangle = \frac{\sum_{n=0}^{\infty} nh\nu e^{-nh\nu/kT}}{\sum_{n=0}^{\infty} e^{-nh\nu/kT}}$$

After some algebra...

$$\langle E \rangle = \frac{h\nu}{(e^{h\nu/kT} - 1)}$$

Previously, we had: $u(\nu) d\nu = \frac{8\pi\nu^2}{c^3} \langle E \rangle d\nu$

Include Planck formula:

$$u(\nu) d\nu = \frac{8\pi h \nu^3}{c^3 (e^{h\nu/kT} - 1)} d\nu$$

Planck distribution function

For small ν set $x = \frac{h\nu}{kT} \ll 1$

Expand $e^x = 1 + x + \frac{x^2}{2!} + \dots$

$$e^x - 1 \approx x$$

$$u(\nu) d\nu \approx \frac{8\pi h \nu^3}{c^3 (\frac{h\nu}{kT})} d\nu \approx \frac{8\pi \nu^2 kT}{c^3} d\nu$$

Rayleigh-Jeans Law!

when ν large,

$$u(\nu) d\nu \approx \frac{8\pi h \nu^3}{c^3 e^{h\nu/kT}} \rightarrow 0 \text{ as } \nu \rightarrow \infty.$$

As expected!

In order to get here, we needed Planck's suggestion

$$E = nh\nu \quad n = 0, 1, 2, \dots$$

Energy is continuous on large scales but on small scales "grainy"...

$$h = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}.$$